

(Old Course)
Faculty Code: 003

Subject Code: 001404

Time : $2\frac{1}{2}$ Hours]

- (1) All questions are compulsory.(2) Q-2 and Q-3 have internal options.
- (3) Figures given in the right side are total marks of that question.
- 1 Answer the following questions:

Instructions: (1)

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[Total Marks: 70

- (1) According to LCAO what is the wave function for H_2^+ ?
- (2) What will be the contribution of 'S' orbital if there are 'n' hybrid orbitals in a set ?
- (3) Define organometallic compounds.
- (4) Complete the reaction; $C_2H_5Li \xrightarrow{\Delta}$? .
- (5) Write Schrodinger equation with Laplacian operator.
- (6) Draw structures of Pyrrole and Pyridine showing α and β carbon in the ring.
- (7) Give the decreasing order of aromaticity of thiophene and furan compared to benzene.
- (8) Complete the reaction; \bigcirc $\xrightarrow{\text{KNO}_3 + \text{Conc.H}_2\text{SO}_4}$?
- (9) Give all functional groups present in EAA.
- (10) Give IUPAC name of Ethyl Aceto Acetate.
- (11) What is zero order reaction?
- (12) Define, "Order of a reaction".
- (13) Give particle size of colloidal solution.
- (14) Give two examples of gas-liquid colloidal system.
- (15) Name the methods used for purification of colloidal sol.

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(16) What is the use of fluxing agents in ceramics.

- (17) Name raw materials used for making ceramics.
- (18) Give name of main types of refractories.
- (19) Give classification of polymers based in inter molecular interaction.
- (20) What is meant by thermoplastics?
- 2 (a) Answer any three of the following:

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- (1) Why furan is the least aromatic?
- (2) Give any two reduction reaction of pyridine.
- (3) Give preparation reaction to form 4-methyl Uracil from EAA.
- (4) Give any two preparation reaction for Organo Aluminium compounds.
- (5) What is LCAO? Define it.
- (6) What is the contribution of 's' orbital towards a wave function of sp hybrid orbitals?
- (b) Answer any three of the following:

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- (1) Prove that the sum of square of coefficient of each component atomic orbital in a wave function is unit.
- (2) Derive φ_{M_0} for H_2^+ by LCAO method.
- (3) Explain reaction of ∞ -alkyl aceto acetate with HNO₂ to form oximino compounds.
- (4) Give nitration of pyrrole, furan and thiophene.
- (5) Define sandwich type organometallic compounds with example.
- (6) Give industrial method for the preparation of ethyl aceto acetate.
- (c) Answer any **two** of the following:

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- (1) Discuss the structure of zeise salt.
- (2) Derive the coefficients of wave functions of Sp^2 hybrid orbitals.
- (3) Give any three synthesis and chemical properties of pyrrole.
- (4) Discuss about the reactivity of active methylene compounds.
- (5) Derive Schrodinger equation with potential energy for H_2 molecule.

- 3 (a) Answer any three of the following:
 6 (1) Explain energy of activation.
 (2) Give hydrolysis method for colloidal sol preparation.
 (3) Which methods are used to apply colours to the pottery?
 - (4) What is meant by basic refractories?
 - (5) Define natural and synthetic polymers.
 - (6) Explain linear polymers.
 - (b) Answer any three of the following:
 - (1) Give reasons for failure of collision theory.
 - (2) Describe classification of polymers based on structure.
 - (3) List the characteristics of good refractories.
 - (4) Explain, "Tyndall Effect".
 - (5) Why inversion of sucrose in presence of dilute HCl is called preudo first order reaction?
 - (6) Determine the unit of rate constant for third order reaction.
 - (c) Answer any **two** of the following:
 - (1) Derive equation of rate constant for the first order reaction and explain half life period.
 - (2) Discuss about the application of colloids.
 - (3) List the methods used to derive the order of reaction and describe graphical method.
 - (4) Discuss in detail; the grinding, mixing and body preparation of ceramics.
 - (5) Write a note on cationic and anionic addition polymerization.